PHOTONUCLEAR METHOD OF PRODUCTION OF FREE ¹⁵³Sm BY USE OF NANOPARTICLES OF SAMARIUM OXIDE AND CLINOTILOLITE

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Nanoparticles of Sm_2O_3 and clinoptilolite were used for obtaining ¹⁵³Sm with a high specific activity by means of using recoil nuclei from reaction ¹⁵⁴Sm(γ ,n)¹⁵³Sm in clinoptilolite. After an irradiating nanoparticle mixture by means of bremsstrahlung of electron accelerator, the Sm₂O₃ removed by dissolution in a hydrochloric acid. The part of recoil atoms of ¹⁵³Sm in clinoptilolite nanoparticles in case use of a bremsstrahlung with the maximum energy of 12.5 MeV is 12.3%. Paths of augmentation of a specific activity and yield ¹⁵³Sm are considered for use of photonuclear reactions.

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INTRODUCTION

All lanthanides have similar chemical properties, they should have similar labeling procedures, and 153 Sm might easily be replaced by other radiolanthanides. An important measure of therapeutic use radiolanthanides is their energy deposition in tumors and in normal tissue. The absorbed dose to normal tissue, and especially to critical organs, needs to be kept as low as possible [1 - 3]. 153 Sm widely used for the palliative treatment of bone metastases.

Metastatic bone disease is the most frequent manifestation of progression of many cancer diseases. The frequency of bone metastasis in breast cancer varies according to different authors, from 47 to 85% of prostate cancer – from 33 to 85%, thyroid – from 28 to 60%. The pain is the most common symptom of the metastatic bone disease. For example, in breast cancer patients with bone metastases, the bone pain is found in 70...80%. In many cases, pain is the dominant factor in worsening the quality of life of the patient. In this context, the deliverance of the pain or, at least, decrease its intensity is one of the most important factors in improving the quality of life of cancer patients with bone metastases. Different methods used during the treatment of patients with the metastatic bone disease: external beam radiation therapy, chemotherapy, hormone therapy, bisphosphonates, surgical treatment, symptomatic medication, radionuclide therapy.

Radiation therapy is a traditional and one of the most effective methods. The method is based on the possibilities of some β -emitting products accumulate in the bone metastases. «Internal» β -irradiation of metastatic tissue allows achieve reduction of tumor infiltration and ensure prolonged analgesia. In the last 20 years, the use of this type of treatment has become common in many radiological and oncology clinics around the world. The use radiopharmaceuticals spectrum is increasing.

Range R₉₀ for β -particles from ¹⁵³Sm in water, with the energy of 810 keV is 1.45 mm. This causes noticeable damage to bone marrow. Ranges R₉₀ of high-energy β -particles of ¹⁶⁹Er, ¹⁷⁵Yb, and ¹⁷⁷Lu in water are 0.6, 0.8, and 0.85 mm, respectively. Therefore, isotopes of ¹⁶⁹Er, ¹⁷⁵Yb, and ¹⁷⁷Lu significantly less are damaging bone marrow. As a result, more acceptable characteristics are for $^{169}\text{Er}, ^{175}\text{Yb}, ^{177}\text{Lu}$, with average energy $\beta\text{-particles 100, 127, and 133 keV, respectively:}$

Decay Data for the ¹⁵³Sm, ¹⁷⁵Yb, and ¹⁶⁹Er

Iso-	Decay	Mean and maximum	E _v , keV
tope	period,	energy β-particles	(intensity,
	hours	(intensity), keV (%)	%)
¹⁵³ Sm	46.44	199.5, 634.7 (31.3),	103.2
		225.3, 704.4 (49.4),	(29.2)
		264.3, 807.6 (18.4)	
¹⁶⁹ Er	223.2	98.3, 342.9 (45),	_
		101.0, 351 (55)	
¹⁷⁵ Yb	100.6	19.0, 73.8 (20.4)	114 (3.9)
		102.4, 356.3 (6.7)	146 (0.67)
		139.9, 470.1 (72.9)	282 (6.1)
			396 (13.2)
¹⁷⁷ Lu	160.8	47.7, 177.0 (11.6)	54.6 (1.6)
		111.7, 385.3 (9.0)	55.8 (2.8)
		149.35, 498.3 (79.4)	113 (6.17)
			208 (10.4)

The production of ¹⁵³Sm is based on the radiative capture of neutrons by isotope ¹⁵²Sm (206 barns). For neutron flux in the reactor $5 \cdot 10^{13}$ n/cm²·s and irradiation during 2 days the ¹⁵³Sm specific activity 345 mCi/mg achieved (one atom ¹⁵³Sm on ¹⁵²Sm 1280 atoms) [4]. The maximum specific activity of ¹⁵³Sm is 436 Ci/mg. However, in the production of ¹⁵³Sm in a nuclear reactor, unavoidable presence ¹⁵²Sm occur, which causes problems when using monoclonal anti-bodies, fragments or small peptides in cancer therapy. Also, ¹⁵³Sm flow into the tumor is limited because of the large amount of the impurity isotope ¹⁵²Sm. This causes problems of palliative therapy of bone metastases with certain primary cancers. Use free ¹⁵³Sm will significantly reduce the side effects of cancer therapy.

¹⁵³Sm free production is possible using charged particles [5]. The yield of thick ¹⁵³Sm target is up to 500 GBq/K using deuterons with the energy of 40 MeV. However the economic characteristics of the production of free ¹⁵³Sm led to the appearance of the project of its production using magnetic separation of ^{152,153}Sm mixture of isotopes, which are obtained, for a reactor [6]. Methods production of free isotopes for palliative treatment of bone metastases were implemented and by means of reactors. For example, for the production of ¹⁷⁷Lu the nuclear reaction ¹⁷⁶Yb(n, γ)¹⁷⁷Yb \rightarrow ¹⁷⁷Lu used, in which the cross-section of radiative thermal neutron capture is only 2.4 bn. The ¹⁷⁶Yb abundance is 12.73%. Therefore, the laser enrichment of ¹⁷⁶Yb [7] is used to increase the yield of free ¹⁷⁷Lu.

The aim of the present article was to investigate the feasibility of producing the free radionuclide ¹⁵³Sm by use of recoil nuclei out of Sm₂O₃ nanoparticles from ¹⁵⁴Sm(γ ,n)¹⁵³Sm reaction.

RESULTS AND DISCUSSION

For the production of 153 Sm with high specific activity was used Szilard-Chalmers method [8]. Nanoparticles Sm₂O₃ and clinoptilolite were used as donor and acceptor, respectively. For the concentration of recoil nuclei in among donor (clinoptilolite), nanoparticle sizes Sm₂O₃, containing an activatable element, must be less than or equal to the range of the recoil nuclei (Fig. 1).



Fig. 1. ¹⁵³Sm ranges in natural samarium oxide

The energy of the recoil nucleus depends on the pulse gamma ray, neutron pulse emitted, the angle between the direction of doing these pulses, where E_r – recoil energy samarium atom, the M is the mass of samarium atom, m – mass of the neutron, the Q – energy nuclear reaction, c – the speed of light, θ – the angle between the directions of the neutron is-started-up and the incident photon, and has the following form:

$$E_{r} = \frac{ME_{\gamma}^{2}}{2(M+m)^{2}c^{2}} + \frac{m^{2}}{2(M+m)^{2}} \left\{ (E_{\gamma} + Q)2(M+m) - \frac{E_{\gamma}^{2}}{c^{2}} \right\} - \frac{E_{\gamma}Mm\cos\theta}{2c(M+m)^{2}} \left\{ \frac{(E_{\gamma} + Q)2(M+m) - \frac{E_{\gamma}^{2}}{c^{2}}}{Mm} \right\}^{1/2}.$$
 (1)

The energy spectrum of neutrons depends on the incident bremsstrahlung, target material. The evaporation model for compound nuclei predicts that the emitted neutron energy distribution approaches the form of a Maxwell distribution [9, 10]:

$$w(E_n) = const \frac{E_n}{\theta^2} \exp(-\frac{E_n}{\theta}),$$

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where $\theta = [(E_{\gamma} - B_n)/a]^{1/2}$, B_n – separation energy of neutron, E_{γ} – bremsstrahlung energy. The constant *a* defines the speed of density ascending of nucleus levels at increasing of energy. The experimental estimate of this constant is $\approx A/15 \text{ MeV}^{-1}$.

The estimate of the average energy of neutrons for a gamma radiation with the energy of 12 MeV of reaction 154 Sm(γ ,n) 153 Sm is equal 628 keV [9, 10]. Therefore, the average energy of recoil nuclei of 153 Sm is equal 4.1 keV. For this energy recoil nuclei, 153 Sm can leave nanoparticles of Sm₂O₃ from a depth of 7.25 mm (see Fig. 1).

The procedure of transforming clinoptilolite in the nanosize state was the following: the grinding of clinoptilolite in an agate mortar for a long time, the precipitation of powder in the distilled water with the subsequent centrifugation. Sm₂O₃ and clinoptilolite samples weighing 300 mg were ground in an agate mortar. After vigorous mechanical stirring Sm₂O₃ and clinoptilolite, sample was irradiated for 3 hours with a maximum bremsstrahlung energy 12.5 MeV. Samarium is being used of natural isotopic distribution. Prevalence of isotopes ¹⁵⁴Sm, ¹⁵²Sm, ¹⁴⁷Sm and ¹⁴⁴Nd is 22.4, 26.6, 15.09, and 3.15%, respectively. The reaction cross section ¹⁵⁴Sm(γ ,n)¹⁵³Sm (T_{1/2}=46.44 hours) at the maximum at 12.5 MeV is 260 mb (Fig. 2).



Fig. 2. Cross section of reaction ${}^{154}Sm(\gamma,n){}^{153}Sm$ [11]

The reaction of 152 Sm(γ ,n) 151 Sm causes the production of 151 Sm isotopes with a half-life of 90 years with low radiation. Reaction 147 Sm(γ ,n) 146 Sm leads to producing 146 Sm isotope with a half-life of 5 \cdot 10⁷ years through alpha decay. Reactions 144 Sm(γ ,n) 143 Sm (T_{1/2}=8.83 min) \rightarrow 143 Pm (T_{1/2}=265 days) is implemented by hours of isotopes with low levels of radiation.

The activity of radioisotopes obtained in reactions $^{154}\text{Sm}(\gamma,n)^{153}\text{Sm}$ in clinoptilolite and solution Sm_2O_3 has been measured by Ge(Li)-detector with volume 50 cm³ and with energy resolution 3.2 keV in the area of 1332 keV (Fig. 3). Only lines of gamma radiation of ^{153}Sm are observed in the spectrum. Note, that recoil nuclei are stopped in the clinoptilolite from all these reactions. Therefore, to obtain 100% of the ^{153}Sm recoil atoms must be used enriched isotope ^{154}Sm .

The sample of Sm_2O_3 and clinoptilolite after an irradiating has been placed into hydrochloric acid (1 ml) for dissolution Sm_2O_3 . After a wash in distilled waters, the rest of clinoptilolite has been subjected centrifugation. Have again added the distilled water in clinoptilolite and have spent centrifugation. After first and second centrifugation the activity of solution Sm_2O_3 was 74.6% and 13.6% of initial, accordingly. The activity clinoptilolite with recoil nuclei of ¹⁵³Sm was 12.3% from the initial.



bremsstrahlung with $E_{max} = 12.5 \text{ MeV}$

The 1 ml of a hydrochloric acid was added in clinoptilolite for excision of Sm_2O_3 sediment again. Then, the distilled water was added in clinoptilolite and was spent a centrifugation. In this case, clinoptilolite activity with recoil nuclei of ¹⁵³Sm was 3.09% from the initial. The estimate of a part of atoms ¹⁵³Sm in clinoptilolite for the average diameter of 150 nanometers equaled 6.4%.

At centrifugation the velocity of subsidence of clinoptilolite particles was being determined out of the equation:

$$V = \frac{3g(\rho - \rho_0)r^2}{9\eta},$$
 (2)

where ρ , ρ_0 – density of clinoptilolite particles and water, accordingly, g – acceleration of free falling, r – particle radius, η – dynamic viscosity of water. The velocity of turnovers of the centrifuge was 12 thousand/s. In this case, the sedimentation at a centrifugation was accomplished for the diameter of clinoptilolite particles from 60 nanometers.

Therefore, the part nanoparticles clinoptilolite is less in the dimension 60 nanometers can get in decant solution Sm_2O_3 . More effective are use nanoparticles Sm_2O_3 as it is possible the smaller dimensioning (\emptyset 10...20 nanometers). It will allow enlarging an amount of recoil nuclei ¹⁵³Sm which will abandon nanoparticles Sm_2O_3 (to 20% for \emptyset 20 nanometers). Diameter nanoparticles clinoptilolite should compound about 100 nanometers [12].

For the mixture of Sm_2O_3 nanoparticles and clinoptilolite on the linear accelerator of electrons of NSC KIPT with the energy 23 MeV and a current 700 μ A is possible to produce 1 Ci ¹⁵³Sm during the day by using of samarium (40 g) with a natural isotopic composition. In the target of similar mass, but enriched in ¹⁵⁴Sm, the daily yield can attain 5 Ci for ¹⁵³Sm.

CONCLUSIONS

The possibility of photonuclear production of free 153 Sm medical radioisotopes produced by the reaction 154 Sm(γ ,n) 153 Sm (T_{1/2}=46.44 hours) by Szilard-Chalmers method was investigated. In this case, the nanoparticle

composition of Sm_2O_3 and clinoptilolite was used. As a result, there is prepared with high specific activity of ¹⁵³Sm which is necessary for systemic radionuclide therapy, especially when using peptides with pharmacological side effects.

In NSC KIPT on the linear accelerator of electrons with E=23 MeV and a current 700 μ A it is possible to produce 1 Ci ¹⁵³Sm during the day by using of samarium oxide nanoparticles (40 g) of natural isotope composition.

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ФОТОЯДЕРНЫЙ МЕТОД ПРОИЗВОДСТВА СВОБОДНОГО ¹⁵³Sm ПРИ ИСПОЛЬЗОВАНИИ НАНОЧАСТИЦ ОКСИДА САМАРИЯ И КЛИНОПТИЛОЛИТА

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Наночастицы Sm_2O_3 и клиноптилолита использовались для получения ¹⁵³Sm с высокой удельной активностью посредством торможения ядер отдачи из реакции ¹⁵⁴Sm(γ ,n)¹⁵³Sm в клиноптилолите. После облучения тормозным излучением электронного ускорителя Sm_2O_3 удалялся растворением в соляной кислоте. Доля атомов отдачи ¹⁵³Sm в наночастицах клиноптилолита при использовании тормозного излучения с максимальной энергией 12,5 MэB составила 12,3%. Обсуждаются пути увеличения удельной активности и выхода ¹⁵³Sm при использовании фотоядерных реакций.

ФОТОЯДЕРНИЙ МЕТОД ВИРОБНИЦТВА ВІЛЬНОГО ¹⁵³Sm ПРИ ВИКОРИСТАННІ НАНОЧАСТИНОК ОКСИДУ САМАРІЮ І КЛИНОПТИЛОЛІТУ

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Наночастинки Sm₂O₃ і клиноптилоліту використовувалися для одержання ¹⁵³Sm з високою питомою активністю за допомогою гальмування ядер віддачі з реакції ¹⁵⁴Sm(γ ,n)¹⁵³Sm у клиноптилоліті. Після опромінення гальмовим випромінюванням електронного прискорювача Sm₂O₃ видалявся розчиненням у соляній кислоті. Доля атомів віддачі ¹⁵³Sm у наночастинках клиноптилоліту при використанні гальмового випромінювання з максимальною енергією 12,5 MeB склала 12,3%. Обговорюються шляхи збільшення питомої активності і виходу ¹⁵³Sm при використанні фотоядерних реакцій.