

THE USE OF NEGATIVE BIAS POTENTIAL FOR STRUCTURAL ENGINEERING OF VACUUM-ARC NITRIDE COATINGS BASED ON HIGH-ENTROPY ALLOYS

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The effect of negative bias potential ($U_b = -40, -110, \text{ and } -200 \text{ V}$) during the deposition of multi-element coatings on their composition, structure and mechanical properties was studied. It was established that during the transition from a multi-element alloy to a nitride, a single-phase state possible to form on its basis (based on the fcc metal lattice, structural type NaCl). In this case, the composition (FeCoNiCuAlCrV)N of coatings with increasing U_b is depleted by the element with the lowest enthalpy of formation of nitride (Cu). In (AlCrTiNbSi)N and (AlCrTiZrNbV)N coatings, the content of low-mass elements (Si and Al) decreases with increasing U_b . In (TiZrHfVNb)N coatings of strong nitride-forming elements with increasing U_b to 200 V, the composition practically does not change. The structure of such coatings is characterized by the presence of a texture with the [111] axis. The presence of weak nitride-forming elements in (FeCoNiCuAlCrV)N coatings leads to the formation of texture [110] for large $U_b = 110 \dots 200 \text{ V}$. In such coatings, the hardness does not exceed 35 GPa. It is shown that to achieve high hardness at high U_b it is necessary to increase the content in the high-entropy alloy of elements with high nitride-forming ability. In this case, in (TiZrHfVNb)N coatings (made of strong nitride-forming elements with a large mass) at $U_b = 200 \text{ V}$, the hardness exceeds 45 GPa.

PACS: 64.75.St, 81.07.Bc, 62.25.-g, 61.05.cp, 61.82.Rx

INTRODUCTION

The traditional approach to creating new materials is to select one element as a basic element. To obtain the desired combination of mechanical and/or other operational properties, the base material is alloyed with additional elements [1, 2]. As basic elements, Fe, Cu, Al, Ni, Mg, etc. are used. However, in recent years, thanks to the use of the structural engineering method (structural design), it has been possible to develop new mechanisms for the formation of composite materials [3–5], expand the possibilities of managing the structural state in non-equilibrium conditions [6] and stabilize metastable phase-structural states [7].

Based on structural engineering, the highest mechanical properties were achieved for coatings obtained by vacuum-plasma methods [8–10]. This is determined by a significant expansion of the possibilities of controlling the structural state in non-equilibrium conditions typical of modern methods for obtaining materials from plasma flows [11, 12]. In vacuum-plasma methods for obtaining materials, super fast thermalization of atoms occurs [13]. This limits the diffusion mobility of atoms and leads to the formation of a disordered solid solution [14]. To a large extent, high mechanical properties are achieved in multi-element coatings as a result of decomposition of supersaturated solid solutions (spinodal type) [15] and ordering [16].

In recent years, structural engineering has been actively used to produce new multicomponent (high-entropy) alloys containing (as a base material) five or more elements in equal (equiatomic) proportions [17, 18]. Such alloys form disordered solid solutions

[19]. Due to this, various mechanisms of their hardening can be used: solid-solution, dispersion, deformation, etc. [20].

Compared with traditional materials, high-entropy alloys (HEA) have significantly higher functional properties. These properties include: high temperature stability [21], high strength [22], high plasticity and fracture toughness [23, 24].

Note that high configurational entropy, caused by mixing five or more elements, contributes to the formation of a disordered solid solution. Unlike traditional alloys, HEA are formed with fairly simple crystal structures (for example, body-centered cubic (BCC), face-centered cubic (FCC) or hexagonal close-packed (HCP) [25, 26]), and not from several intermetallic compounds with complex microstructures.

Another effect characterizing HEA is a strong lattice distortion caused by a large difference in the atoms size in a unit cell. Lattice distortion is commonly used to explain the observed hardening of a multicomponent phase compared to binary or ternary systems. To quantify the lattice distortion, the δ parameter is commonly used. It represents the average deviation of the metal radii of the elements from the average radius [27]:

$$\delta = \sqrt{\sum_{i=1}^n c_i \left(1 - \frac{r_i}{\bar{r}}\right)^2},$$

where c_i is the atomic percentage of the i -th element; r_i and \bar{r} are the radius of the atom and the average atomic radius.

There is also growing interest in composite materials based on HEA alloys with added p-elements such as C, O or N. This leads to the formation of high-entropy carbides (HEC) [28–30], high-entropy oxides (HEOs) [31, 32] and high-entropy nitrides (HENs) [33]. For the latter class (HENs), a large number of interesting properties have been identified, such as high hardness [34, 35], thermal stability and radiation resistance [36], high wear resistance [37], and corrosion resistance [38, 39].

Unlike HEA, most HENs publications are related to coating materials. This is due to the simplicity of the synthesis of nitrides in the form of coatings deposited in a nitrogen atmosphere. Among the methods of obtaining the most frequently used method is high-frequency magnetron sputtering in a mixed medium of Ar and N₂, with such a parameter as $R_N = N_2/(Ar + N_2)$ [40]. It has been shown that metals that form stable binary nitrides tend to form HENs [41].

At the same time, an important factor that significantly affects the efficiency of nitride formation is the energy of the particles during the formation of nitrides. However, during magnetron sputtering, the degree of ionization is not large. This leads to a low efficiency of influence on the energy characteristics of atoms of such an important parameter as the bias potential of U_b . The use of the vacuum-arc method makes it possible to achieve a high degree of ionization and thereby effectively use U_b to change the energy of the particles.

Therefore, the aim was to compare the effect of U_b for vacuum-arc coatings obtained on the basis of different systems of HEA:

- from elements with relatively low affinity for nitrogen (FeCoNiCuAlCrV);
- when partial replacement of elements with low enthalpy of nitrides formation ΔH_{298}^0 on elements with a greater enthalpy of formation of nitrides AlCrTiZrNbV (with replacement Fe, Co, Ni, Cu on Ti, Zr, Nb), and AlCrTiNbSi system (by introducing Si with higher affinity to the nitrogen);
- consisting of elements with a high affinity for nitrogen (TiZrHfVNb).

SAMPLES AND RESEARCH METHODS

The coatings were deposited by the vacuum-arc method on the modernized installation “Bulat-6”.

Previously manufactured a cathode of the required composition by vacuum-arc remelting of a multicomponent mixture of powders of pure metals. Metals with a purity of at least 99.9% were used as the starting components.

After ten times remelting, the HEA crystallized at a rate of 20 K/s.

Several types of ingots-billets were prepared based on: AlCrVFeCoNiCu, AlCrTiNbSi, AlCrTiZrNbV, and TiZrNbVHf. From these ingots-billets in the future after mechanical processing cathodes were made (typical view shown in Fig. 1).



Fig. 1. General view of the cathode before evaporation

Samples of the size (15x15x2.5 mm) of 12X18H9T steel ($R_a = 0.09 \mu\text{m}$) were chosen as substrates for the deposition of coatings. In the process of deposition, a constant negative bias potential $U_b = -40, -110, \text{ and } -200 \text{ V}$ was supplied to the substrate. The coatings were deposited at a pressure $P_N = 5 \cdot 10^{-3} \text{ Torr}$. The deposition time was about 1 hour, which made it possible to obtain coatings with a thickness of 7...8 μm .

The study of the morphology of the cross section of multi-element coatings was carried out on a JEOL JSM840 scanning electron microscope. The elemental composition of the coatings was studied by analyzing the spectra of the characteristic X-ray radiation generated by the electron beam in a scanning electron microscope.

The phase-structural state was studied on a DRON-4 diffractometer in Cu-K α radiation. For monochromatization a graphite monochromator was used, which was installed in the secondary beam (in front of the detector) [42]. Measurement was carried out in the angular range of $2\theta = 20 \dots 80^\circ$. All diffraction peaks from the planes with the highest reticular density of atoms fall into this angular range. Scan step $\theta = 0.02^\circ$. Microindentation was carried out on the installation “Micron-gamma” [43] with a load up to $F = 0.5 \text{ N}$ using a Berkovich diamond pyramid with a sharpening angle of 65° , with automatically performed loading and unloading for 30 s.

RESEARCH RESULTS

In multi-element coatings, the composition may vary significantly with varying process parameters. In this work, U_b used as a technological parameter affecting on the composition.

First, we consider the effect of U_b on the composition of coatings, which consist of elements with the lowest enthalpy of formation of nitrides (based on AlCrVFeCoNiCu).

In the initial state, the cathode for evaporation had the elemental composition: Fe – 15, Co – 15, Ni – 13, Cu – 20, Al – 8, Cr – 19, V – 10 at.%. The elemental composition of the deposited coatings are shown in Tabl. 1.

Table 1
The elemental composition (FeCoNiCuAlCrV)N coatings (at.%) taking into account the nitrogen content

U_b , V	Ti	Cr	Al	Nb	Si	N
-40	21.0	12.9	10.6	7.5	2.7	45.3
-110	23.1	13.5	10.3	7.2	1.9	44.0
-200	30.1	13.4	9.8	7.5	1.3	37.9

From Tabl. 1 it can be seen that the supply of U_b leads to a selective change in composition. To the greatest extent the content of Cu decreases. Also significantly decreases the content of N. The relative content of the remaining elements slightly increases.

In coatings obtained on the basis of the AlCrTiZrNbV alloy (in which, instead of Cu, Fe, Co, Ni, introduce elements Ti, Zr, Nb with strongly bond with nitrogen), a change in U_b also leads to a change in the elemental composition. The results of the elemental analysis are given in Tabl. 2.

It can be seen that with this combination of d-elements in a HEA, the critical element by which occurs of depleted the alloy is Al. The observed relative decrease in the aluminum content is apparently determined by its selective spraying of heavy atoms from the growth surface. As U_b increases, the N content in the coating also decreases. However, this decrease is less significant compared to (FeCoNiCuAlCrV)N coatings.

Table 2
Elemental composition (AlCrTiZrNbV)N coatings obtained at different bias potential

U_b , V	Fe	Co	Ni	Cu	Al	Cr	V	N
-40	10.8	10.3	8.8	12.8	5.8	13.4	7.2	30.8
-110	12.3	12.3	10.6	7.6	5.5	13.9	8.6	29.0
-200	14.0	13.0	12.6	6.8	6.3	15.8	9.0	21.5

Tabl. 3 shows the resulting data on the influence of U_b on the composition of coatings based on the AlCrTiNbSi high-entropy alloy. As the main difference of the elemental composition from the previous series can be considered the addition of a strong nitride-forming element – Si. However, with increasing U_b , occurs a decrease in the content of Si. This gives reason to believe that the main reason for this effect is the sputtering of light atoms (as the predominant factor in the formation of the composition of coatings in the deposition process) in comparison with nitride-forming.

Table 3
The elemental composition (AlCrTiNbSi)N coatings of 3 series obtained at different bias potential

U_b , V	Ti	Zr	Al	V	Nb	Cr	N
-40	11.8	10.0	6.4	9.1	12.1	10.3	40.3
-110	12.4	11.6	4.3	9.2	14.7	11.7	36.1
-200	11.3	15.4	2.8	11.0	18.7	16.3	24.6

To the greatest extent of all strong nitride-forming elements have been used in coatings on based TiZrNbVHf. Tabl. 4 shows the results of elemental analysis for different deposition conditions. It can be seen that in this case (when a large specific content of heavy metal atoms in the alloy) during coating deposition, occurs the relative content of atoms V decreases (i. e. the element with the lowest relative heat of formation of nitrides). At the same time, the nitrogen content in coatings for the whole range of values $U_b = -40 \dots -200$ V remains relatively large (close to stoichiometric for mononitride).

Table 4
Elemental composition (at.%) of coatings based on HEA TiZrNbVHf

U_b , V	Ti	Zr	Hf	Nb	V	N
-50	19.8	8.3	10.1	6.4	5.1	50.3
-100	21.2	7.9	9.8	6.1	4.9	49.8
-200	18.2	11.9	10.5	6.9	3.2	48.3

Thus, in the case of a large difference in the masses of the metal atoms that make up the HEA, with an increase in the particle energy (with an increase in U_b), occurs a predominant sputtering of light metal atoms. For compositions of heavy metal atoms, occurs predominant depletion of coatings with atoms having the lowest bond energy with nitrogen.

To establish the laws of the influence of U_b on the morphology of growth of coatings, studies of the cross section during the fracture of the “coating on copper foil” system were conducted.

It is known that the main drawback of nitride coatings obtained by the vacuum-arc method is the formation of atoms not reacting with nitrogen in the form of a droplet phase. As can be seen from Fig. 2, with an increase in U_b in HEA, occurs decrease both the average number and the size of droplet formations.

To research the phase composition and structural state of the coatings, an X-ray diffraction analysis method was used. Common to all types of coatings is the formation of a single-phase state solid solution based on the FCC metal lattice (structural type NaCl).

Fig. 3 shows the XRD patterns of (AlCrVFeCoNiCu)N coatings obtained at pressure $P_N = 5 \cdot 10^{-3}$ Torr and $U_b = -40$ V (see Fig. 3, spectrum 1) and $U_b = -200$ V (see Fig. 3, spectrum 2).

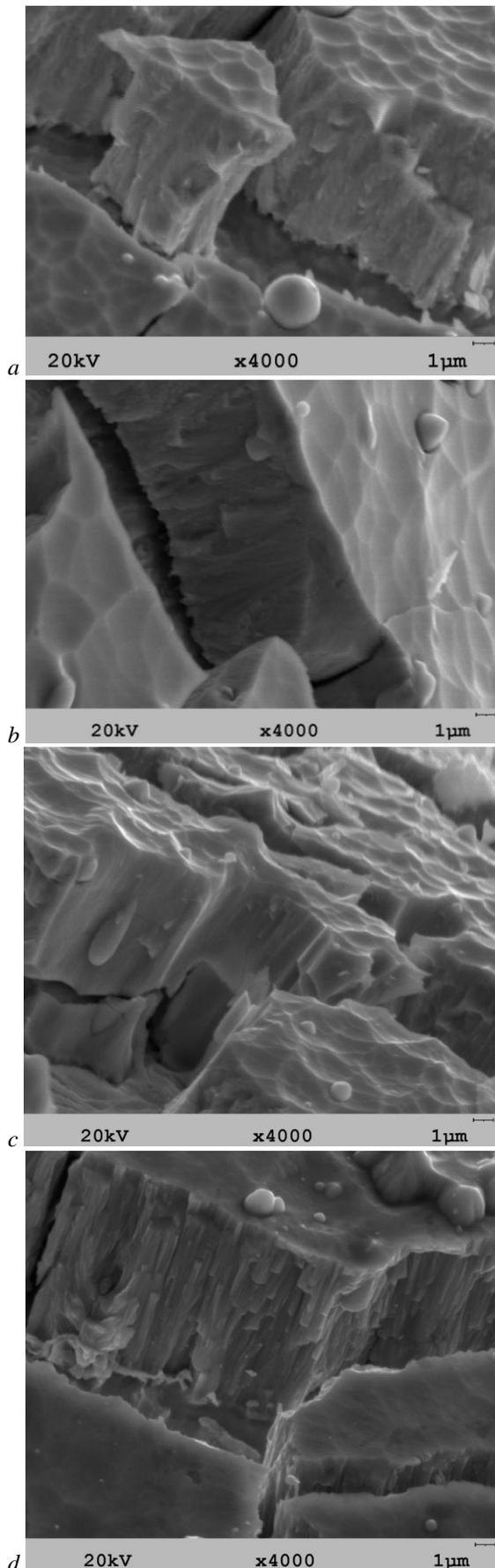


Fig. 2. Morphology of the side surface (TiZrHfVNb)*N* coatings deposited at $U_b = -40$ (a) and -200 V (b) and (TiZrAlVNbCr)*N* coatings deposited at $U_b = -110$ (c) and -200 V (d)

An analysis of the diffraction spectra shows that at the smallest $U_b = -40$ V a polycrystalline structure is formed with a preferential orientation of a part of crystallites with the [111] axis perpendicular to the growth surface (texture of the axial type with low perfection). With an increase in U_b to -200 V, a change in the type of texture is observed. In this case, the axis of the predominant growth of crystallites becomes [110] (see Fig. 3, spectrum 2).

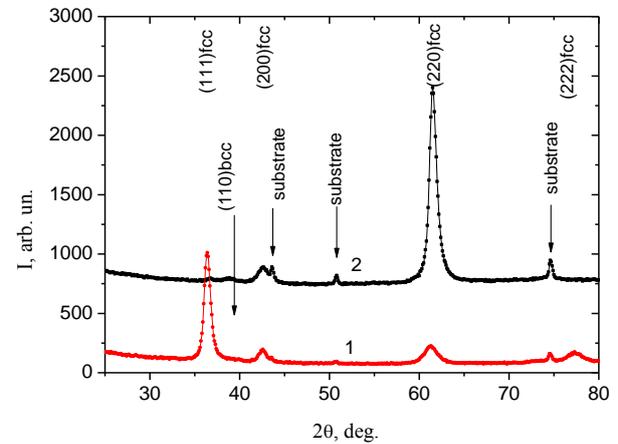


Fig. 3. XRD patterns of the (AlCrVFeCoNiCu)*N* coatings obtained at $P_N = 5 \cdot 10^{-3}$ Torr: $U_b = -40$ (1), -200 V (2)

Important structural characteristics of the coatings are the average crystallite size. In coatings of this series, the average crystallite size is about 11 nm.

XRD patterns of (AlCrTiZrNbV)*N* high-entropy alloy coatings are shown in Fig. 4. It is seen that in this case, the change in U_b to a lesser extent affects the change in the preferential orientation of growth. At $U_b = -40$ V, the coatings are practically non-textured (see Fig. 4, spectrum 1). At $U_b = -110$ V, texture formation with the [100] axis is visible (see Fig. 4, spectrum 2). However, with a larger U_b , this type of texture does not appear. With an increase in U_b to -200 V, texture formation with (a low perfection) [110] axis is seen (see Fig. 4, spectrum 3).

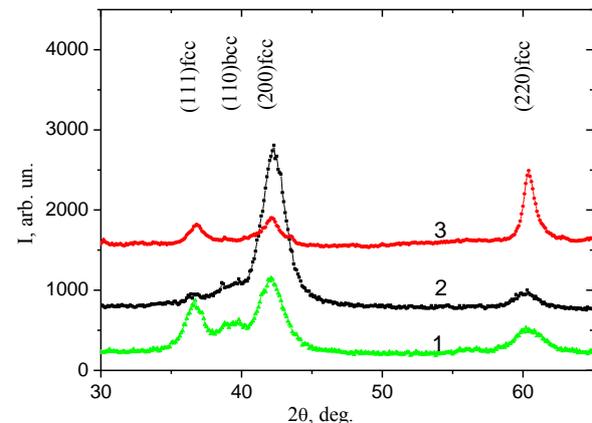


Fig. 4. XRD patterns of the (AlCrTiZrNbV)*N* coatings obtained at $P_N = 5 \cdot 10^{-3}$ Torr: $U_b = -40$ (1); -110 (2); -200 V (3)

The average crystallite size for coatings obtained at $U_b = -40 \dots -110$ V was about 5 nm, which is significantly less than the typical crystallite size for (AlCrVFeCoNiCu)*N* coatings. In coatings obtained at $U_b = -200$ V,

the average crystallite size increased strongly and was about 10.5 nm. The latter effect, apparently, is a consequence of the action of defects annealing in the process of ion bombardment and the relaxation processes associated with it.

It should be noted that for similar types of coatings, but produced under conditions of low plasma ionization (magnetron sputtering), results were obtained in [44]. In this case, it was shown that as U_b changes from 0 to -150 V, the structural state of the coatings changes from texture with the [100] axis to texture with the [111] axis. At the same time, the compressive macrostresses increase from -0.9 to -6.6 GPa. Also, at highest U_b , the highest hardness was 36.9 GPa and the adhesive strength was 60.7 N.

For multi-element nitride (AlCrTiNbSi)N diffraction spectra are shown in Fig. 5. As can be seen from the diffraction spectra, the main phase is a nitride of a solid solution with an FCC type crystal lattice. As the bias potential increases from -40 to -200 V (spectra 1 and 3, respectively), the preferential orientation of growing crystallites changes from the [111] texture axis to [100], which is typical when the material is depleted on the interstitial element (N) [12].

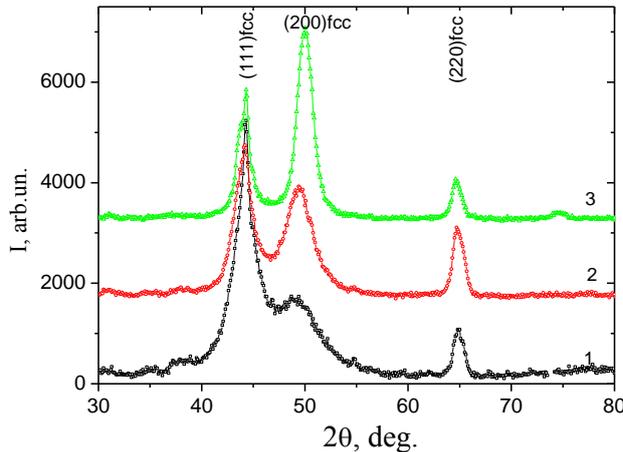


Fig. 5. XRD patterns of the (AlCrTiNbSi)N coatings: $U_b = -40$ (1), -110 (2), -200 V (3)

The lattice period decreases from 0.371 to 0.363 nm with an increase in the bias potential. The crystallite size in the direction perpendicular to the surface plane increases from 3.5 to 8.1 nm with an U_b increase.

XRD patterns of the (TiZrHfVNb)N coatings are shown in Fig. 6. It can be seen that the [111] texture of low perfection is formed at the lowest potential of -50 V (Fig. 6, spectrum 1).

At a higher negative bias potential of -100 V, the tendency to form coatings with the [111] texture remains (see Fig. 6, spectrum 2). In this case, the degree of perfection of the texture (compared with $U_b = -50$ V) is much higher, which is manifested in the almost complete absence of diffraction peaks in the spectrum from other planes.

The highest degree of texturing corresponds to the coatings obtained at the highest $U_b = -200$ V (see Fig. 6, spectrum 3).

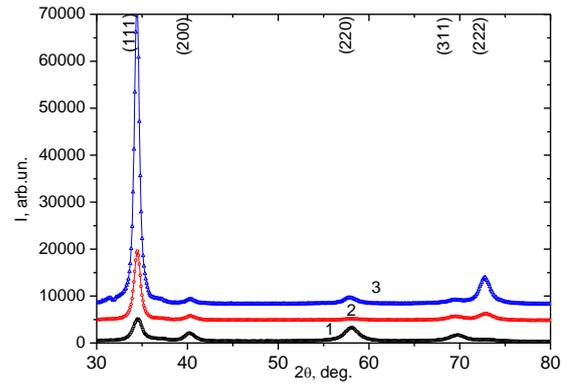


Fig. 6. XRD patterns of the (TiZrHfVNb)N coatings: $U_b = -50$ (1), -100 (2), -200 V (3)

The crystallite size with increasing U_b varied from 8.5 nm (-50 V) to 11.7 nm (-200 V).

The final stage of structural engineering is the establishment of patterns of influence of structural states on functional (physico-mechanical) properties. The most expressive and versatile is the study of physico-mechanical properties by measuring hardness.

Fig. 7 shows the comparative dependences of $H(U_b)$ for all the coatings studied in the work. It can be seen that the obtained dependences $H(U_b)$ can be divided into 3 characteristic types of curves.

The first type can be attributed to the form of a curve for nitride HEA coatings from elements having a relatively low nitride-forming ability (FeCoNiCuAlCrV). For this type of $H(U_b)$ dependence, the highest hardness is achieved at the lowest U_b (see Fig. 7, dependence 1). With an increase in U_b , occurs a sharp decrease in the hardness of coatings of this type.

The second type includes coatings in which metal elements are present with both high and relatively low heat of formation of nitrides. This type of $H(U_b)$ dependence is characteristic of the systems (AlCrTiZrNbV)N (see Fig. 7, dependence 2) and (AlCrTiNbSi)N (see Fig. 7, dependence 3). In this case, the highest hardness is achieved with average values of U_b .

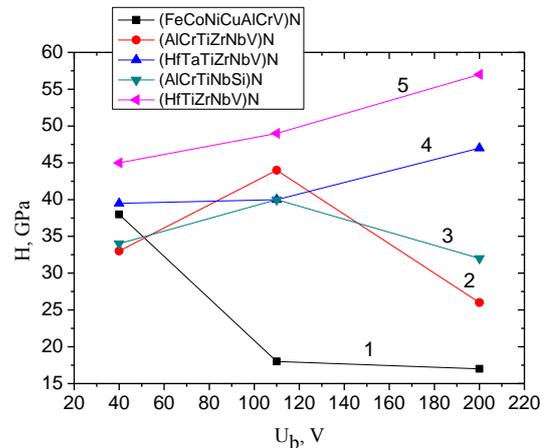


Fig. 7. Dependence of hardness (H) of the magnitude of the bias potential (U_b) for vacuum-arc coatings based on nitrides of HEA:

- 1 – (FeCoNiCuAlCrV)N; 2 – (AlCrTiZrNbV)N;
3 – (AlCrTiNbSi)N; 4 – (HfTaTiZrNbV)N;
5 – (HfTiZrNbV)N

The type of curve for a HENs coatings consisting of elements with the highest nitride-forming ability can be attributed to the third type. This type includes (HfTiZrNbV)N coatings (see Fig. 7, dependence 5), as well as described previously in [45] the high-entropy coatings (HfTaTiZrNbV)N (see Fig. 7, dependence 4). It is seen that this type of coating is characterized by a continuous increase in hardness with increasing U_b in the range of 40...200 V. In absolute values, the hardness for this type of coating is greatest.

DISCUSSION OF THE RESULTS

During the formation of coatings in a nitrogen atmosphere, the most important thermodynamic characteristic is the enthalpy of formation of nitride ΔH_{298}^0 . For multi-element coatings, this thermodynamic parameter is determined by the contribution of each of the elements. Values ΔH_{298}^0 for nitrides of elements used in this work are given in Tabl. 5.

Table 5

The enthalpy of formation of nitrides for the elements included in the basis of high-entropy alloys

Nitride	Si ₃ N ₄	HfN	ZrN	AlN	TiN	VN	TaN
ΔH_{298}^0 , kJ/mol	748	369.44	365.51	320.29	336.62	251.21	247.02
Nitride	NbN	CrN	MoN	FeN	NiN	CoN	Cu ₃ N
ΔH_{298}^0 , kJ/mol	237.81	118.07	69.50	3.77	-0.84	-8.37	-74.53

Based on the data in Tabl. 5, the elemental composition of HEA with the highest properties may have lower functional characteristics during the formation of nitrides. An example of this is the AlCrVFeCoNiCu alloy studied in this work.

When analyzing the elemental composition, it should be noted that for coatings based on AlCrVFeCoNiCu (containing d-elements with the lowest heat of formation of nitrides) with increasing bias potential in the coating, the content of Fe, Co, Ni, V atoms is increased. Content of the Cu (to the greatest extent), Cr, and Al decreases. Thus, during the deposition of coatings in a nitrogen atmosphere, the relative content of nitride-forming elements increases. At the same time, the relative content of copper atoms (having a low bond energy with nitrogen) is greatly reduced. Thus, the most significant factor in the formation of the elemental composition is the bond strength between the metallic atoms and the nitrogen atoms (in the formation of nitride). This fact indicates that elements with different masses, but high binding energy with nitrogen are retained in the coating. At the same time, elements with a high mass but low enthalpy of formation of nitrides are removed from the coating. This can only happen when spraying near the growth surface, when the bound metal-nitrogen complexes have a low sputtering coefficient.

The change in elemental composition for HENs based on AlCrTiNbSi and AlCrTiZrNbV showed that the use in the first alloy of a strong nitride-forming, but light element (Si) is not an obstacle to reducing its content in the coating with increasing U_b (see Tabl. 3). There is also a depletion of the lightest element (Al) during the deposition of nitride coatings based on the HEA AlCrTiZrNbV (see Tabl. 2).

In coatings based on the heaviest elements (HfTiZrNbV) with an increase in U_b , depletion of the lightest elements (Ti and V) occurs. However, to the greatest extent depletion is V occurs which has a relatively low nitride formation enthalpy compared to Ti. Thus, in this case, 2 critical mechanisms of depletion are manifested: the smallest mass and relatively low ΔH_{298}^0 .

There is also a characteristic decrease in the relative content of nitrogen atoms in the coating with a decrease

in the specific component of the elements with a relatively large ΔH_{298}^0 formation of nitride.

The composition of coatings affects the change in the structural state. For coatings consisting of elements with a relatively small (first series based on AlCrVFeCoNiCu), when applying $U_b = -200$ V, occurs formed a texture with the [110] axis (see Fig. 3, spectrum 2). The formation of such a texture is stimulated by ion bombardment processes and is facilitated by the low content of nitrogen atoms in the coating (providing covalent polar bonds). For coatings based on the HEA AlCrTiZrNbV, similar effect is less pronounced (see Fig. 4). Apparently this determines greater hardness of (AlCrTiZrNbV)N coatings compared with hardness of (AlCrVFeCoNiCu)N coatings at $U_b = -200$ V.

The use of Si as a component leads to a strong dispersion of crystallites (up to 3.5 nm) with a silicon content of 1.9...2.7 at.%. When the silicon content decreases to 1.3 at.%, occurs the average crystallite size increases and the hardness of the coatings decreases.

For coatings consisting of elements with relatively large ΔH_{298}^0 (with a nitrogen composition close to stoichiometric), a texture with an [111] axis perpendicular to the growth surface is formed in the entire range of U_b used (see Fig. 6). In this case, the arrangement of the (111) planes with the highest reticular density parallel to the growth surface provides the minimum energy of the deformation factor ($E_s \sim E\varepsilon^2$). The kinetic factor is also minimized due to the highest growth rate in the [111] direction (by alternating layers of metal and nitrogen atoms with form a multilayer system with an angstromy period in the direction of the [111] axis).

When analyzing the results on the hardness of coatings, it should be noted that despite the fact that the coatings obtained on the basis of an alloy of seven elements (FeCoNiCuAlCrV systems) represent a single-phase state, the hardness of such a coating does not exceed 38 GPa (see Fig. 7). Thus, an analysis of the mechanical properties of coatings from the multicomponent FeCoNiCuAlCrV alloy (in which the number of nitride-forming elements with a strong bond is less than that with a weak bond) showed that such compositions can have high functional properties only at low energy

of bombarding particles (which is accompanied by sputtering of the of the coating during its formation) during deposition. In this paper, this is manifested at the lowest $U_b = -40$ V.

From the comparative data in Fig. 7 it can be seen that for coatings based on (FeCoNiCuAlCrV), the highest hardness was obtained at $U_b = -40$ V. In this case, a polycrystalline practically non-textured structure is formed, which is characterized by obstruction of sliding along the planes of the crystallites (in this case, the material properties become close to the elastic limit). In this case, the level of the maximum achievable hardness is approximately the same for all types of coatings.

At a higher bias potential ($U_b = -110$ V), occurs a relative increase in the hardness of coatings consisting of a stronger nitride-forming elements. And with the largest $U_b = -200$ V, the highest hardness is achieved in (HfTiZrNbV)N coatings consisting of the strongest nitride-forming elements.

Thus, only in multi-element coatings consisting of elements with a high nitride-forming ability (greater gain in free energy during the formation of nitride), an increase in the energy of the particles (due to an increase in U_b) can significantly increase the hardness.

CONCLUSIONS

1. For all types of HEA studied in the work, the deposition in nitrogen atmosphere leads to the formation of a single-phase state (fcc metal lattice, structural type NaCl).

2. It has been established that the selective change in composition during deposition is due to two main processes: if an element with a very low enthalpy of formation of nitride (Cu) is present in an alloy, the coating is depleted of this element, but at a sufficiently high enthalpy of all forming elements, the relative content of the lightest component decreases.

3. An increase in U_b leads to a decrease in the relative content of nitrogen atoms in the coating. This is most pronounced in HENs based on (FeCoNiCuAlCrV). Those coatings with a high specific content of elements having a low affinity with nitrogen.

4. When the content of nitrogen atoms is close to stoichiometric (for mononitride), the axis of the preferential orientation of the [111] crystallites is formed. In the (FeCoNiCuAlCrV)N and (AlCrTiZrNbV)N coatings at $U_b = -200$ V and a large deficiency in nitrogen atoms, a texture with the [110] axis forms. The formation of the [110] texture is determined by the minimal influences of the factor of ion bombardment during the growth of the coating.

5. When classifying the $H(U_b)$ dependencies, 3 characteristic types were revealed. The first type – with the highest H value at low U_b . This type is characteristic of (FeCoNiCuAlCrV)N coatings in which a significant portion of the metal atoms has a small affinity with the nitrogen atoms. The second type of dependence $H(U_b)$ has a curve with a maximum hardness at $U_b \approx -110$ V. This type is characteristic of (AlCrTiZrNbV)N and (AlCrTiNbSi)N coatings. For coatings (HfTiZrNbV)N, consisting of strong nitride-forming elements with increasing U_b , occurs continuous increase in hardness. This is the third type of dependence $H(U_b)$ and in this case the greatest hardness is reached, exceeding 45 GPa.

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Article received 26.11.2018

ИСПОЛЬЗОВАНИЕ ОТРИЦАТЕЛЬНОГО ПОТЕНЦИАЛА СМЕЩЕНИЯ ДЛЯ СТРУКТУРНОЙ ИНЖЕНЕРИИ ВАКУУМНО-ДУГОВЫХ НИТРИДНЫХ ПОКРЫТИЙ НА ОСНОВЕ ВЫСОКОЭНТРОПИЙНЫХ СПЛАВОВ

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Исследовано влияние отрицательного потенциала смещения ($U_b = -40, -110$ и -200 В) при осаждении многоэлементных покрытий на их состав, структуру и механические свойства. Установлено, что при переходе от многоэлементного сплава к нитриду на его основе возможно формирование однофазного состояния (на основе ГЦК металлической решетки – структурный тип NaCl). При этом состав (FeCoNiCuAlCrV)N-покрытий с увеличением U_b обедняется элементом с наименьшей энтальпией образования нитрида (Cu). В (AlCrTiNbSi)N- и (AlCrTiZrNbV)N-покрытиях с увеличением U_b уменьшается содержание элементов с малой массой (Si и Al). В (TiZrHfVNb)N-покрытиях из сильных нитридообразующих элементов с большой массой при увеличении U_b до 200 В состав практически не изменяется. Структура таких покрытий характеризуется наличием текстуры с осью [111]. Наличие слабых нитридообразующих элементов в (FeCoNiCuAlCrV)N-покрытиях приводит при больших $U_b = 110...200$ В к формированию текстуры [110]. В таких покрытиях твердость не превышает 35 ГПа. Показано, что для достижения высокой твердости при больших U_b необходимо увеличивать содержание в высокоэнтропийном сплаве элементов с высокой нитридообразующей способностью. При этом в (TiZrHfVNb)N-покрытиях (из сильных нитридообразующих элементов с большой массой) при $U_b = 200$ В твердость превышает 45 ГПа.

ВИКОРИСТАННЯ НЕГАТИВНОГО ПОТЕНЦІАЛУ ЗМІЩЕННЯ ДЛЯ СТРУКТУРНОЇ ІНЖЕНЕРІЇ ВАКУУМНО-ДУГОВИХ НІТРИДНИХ ПОКРИТТІВ НА ОСНОВІ ВИСОКОЕНТРОПІЙНИХ СПЛАВІВ

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Досліджено вплив негативного потенціалу зміщення ($U_b = -40, -110$ і -200 В) при осадженні багатоелементних покриттів на їх склад, структуру і механічні властивості. Встановлено, що при переході від багатоелементного сплаву до нитриду на його основі можливе формування однофазного стану (на основі ГЦК металеві решітки – структурний тип NaCl). При цьому склад (FeCoNiCuAlCrV)N-покриттів зі збільшенням U_b збіднюється елементом з найменшою ентальпією утворення нитриду (Cu). У (AlCrTiNbSi)N- і (AlCrTiZrNbV)N-покриттях зі збільшенням U_b зменшується вміст елементів з малою масою (Si і Al). У (TiZrHfVNb)N-покриттях, що складаються з сильних нітридоутворюючих елементів з великою масою, при збільшенні U_b до 200 В склад практично не змінюється. Структура таких покриттів характеризується наявністю текстури з віссю [111]. Наявність слабких нітридоутворюючих елементів у (FeCoNiCuAlCrV)N-покриттях призводить при великих $U_b = 110...200$ В до формування текстури [110]. У таких покриттях твердість не перевищує 35 ГПа. Показано, що для досягнення високої твердості при великих U_b необхідно збільшувати вміст у високоентропійному сплаві елементів з високою нітридоутворюючою здатністю. При цьому в (TiZrHfVNb)N-покриттях (з сильних нітридоутворюючих елементів з великою масою) при $U_b = 200$ В твердість перевищує 45 ГПа.